

**A STUDY OF THE ELECTRONIC STRUCTURE OF  
POLYVINYLSILOXANE (CH<sub>2</sub>CHSiO<sub>1.5</sub>)<sub>n</sub> BY X-RAY  
PHOTOELECTRON SPECTROSCOPY AND QUANTUM  
CHEMICAL MODELING IN THE DFT APPROXIMATION**

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The electronic structure of polyvinylsiloxane polymeric chains (Si<sub>2</sub>O<sub>3</sub>(CHCH<sub>2</sub>)<sub>2</sub>)<sub>n</sub> is studied by X-ray photoelectron spectroscopy and quantum chemistry in the DFT approximation. The binding energy of C and O 1s electrons occupying inequivalent positions in the polymer coincides within the experimental accuracy. The binding energies for C and O (284.9 eV and 532.4 eV) and for Si2p-electrons (102.7 eV) well agree with the values for related compounds. The experimental data for the binding energy are reproduced in HF and DFT calculations only with the extended 6-311\*\*(*d*) basis set. The highest occupied levels of the polymer are  $\pi$  orbitals of vinyl groups.

**Keywords:** polymers, siloxanes, XPS, DFT, electronic structure.