

# **$\beta$ -Hydroxyalkylation of Sterically Hindered Phenols with Epoxides in Acid Medium**

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**Abstract**—Reactions of 2,6-dialkylphenols with ethylene oxide, propylene oxide and epichlorohydrin in the presence of  $\text{SnCl}_4$  at the temperature from  $-5$  to  $+5^\circ\text{C}$  leads to the formation of respective phenols containing a hydroxy group in the  $\beta$ -position of the aliphatic chain of the *para*-substituent. The conditions for maximum selectivity of the reaction of 2,6-di-*tert*-butylphenol with ethylene oxide were determined. By HPLC–MS method the directions of the side reactions were explored. The method has been successfully tested in a pilot installation. With 2,6-dimethylphenol instead of 2,6-di-*tert*-butylphenol a sharp increase occurs in the content of ethers in the reaction product. With epichlorohydrin, 2,6-di-*tert*-butylphenol affords a product, which is easily converted into an epoxide containing a sterically hindered phenol in its structure.

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